

# Does The Mpemba Effect really happen?

## Introduction

The Mpemba effect is the unexplained phenomenon that hot water freezes faster than cold water.

## Theory and Experiment "Reproduction of the Mpemba effect"

- Outline:
  - Define "freezing" as "reaching  $-10^{\circ}\text{C}$ ".
  - Freeze  $5^{\circ}\text{C}$  and  $35^{\circ}\text{C}$  water simultaneously and measure the time to reach  $-10^{\circ}\text{C}$ .

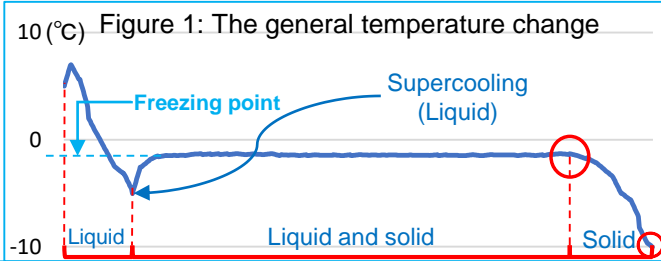
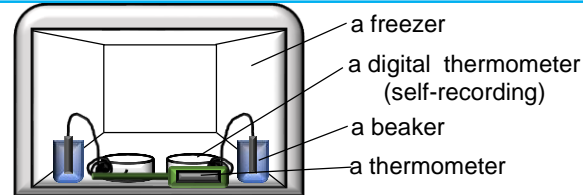


Figure 2: The device to measure the temperature of the central part of liquid

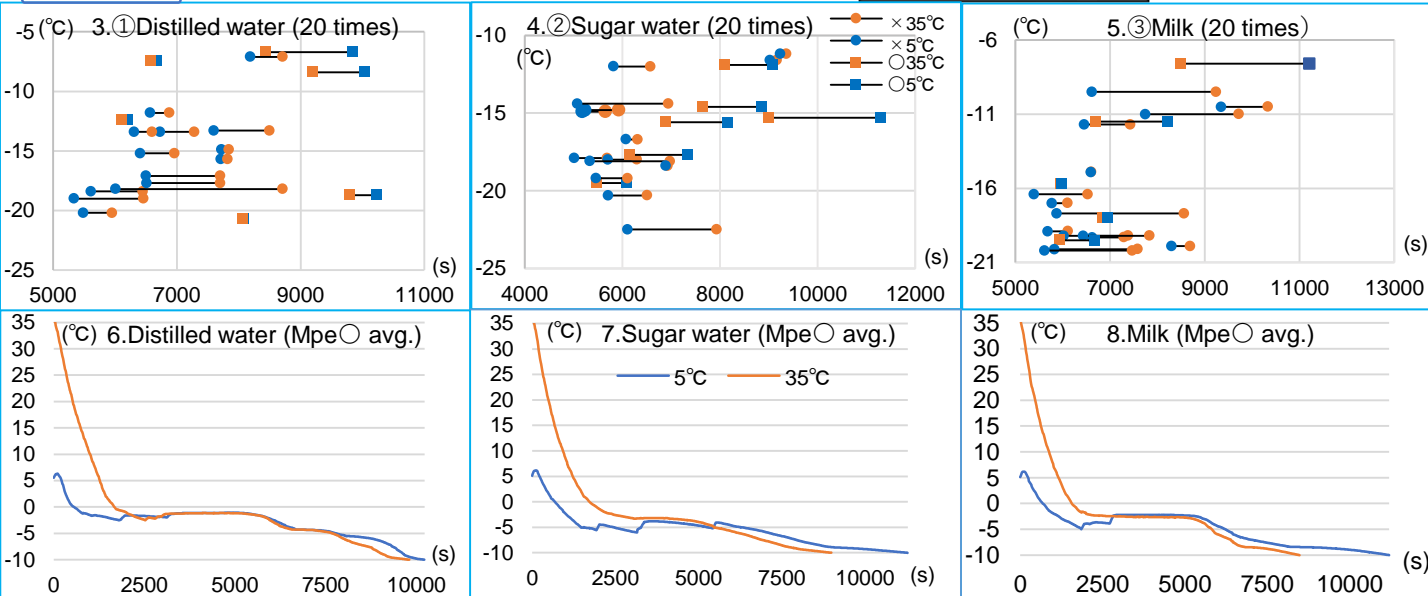
### <The method>

- Put two beakers each of which is poured  $5^{\circ}\text{C}$  and  $35^{\circ}\text{C}$  distilled water, sugar water, and milk (40ml) into a freezer.
- Measure the elapsed time to reach  $-10^{\circ}\text{C}$



## Results

The vertical axis: the initial temperature in the freezer  
The horizontal axis: the elapsed time to reach  $-10^{\circ}\text{C}$



## Discussion

- The probability of the Mpemba effect
  - ① 30% (6/20) ② 30% (6/20) ③ 25% (5/20)
- The correlation between the initial temperature in the freezer and the difference in each elapsed time
  - ① -0.51 ② -0.28 ③ -0.25
- $35^{\circ}\text{C}$  liquid is cooled below freezing point faster than  $5^{\circ}\text{C}$  liquid.
- Regardless of the any kinds of liquid, the higher the initial temperature in the freezer is, the smaller the difference in each elapsed time is.

## Conclusion

The Mpemba effect really happens, and it is a stochastic phenomenon.  
I want to study about another condition under which it is likely to happen from now on.

## References

- <https://kimika.net/rr4reikyakukyokusen.html>  
"What is the cooling curve?"